

## Addition of Difluorocarbene to Steroidal Unsaturated Ketones: Formation of Novel Difluoromethyl Ethers

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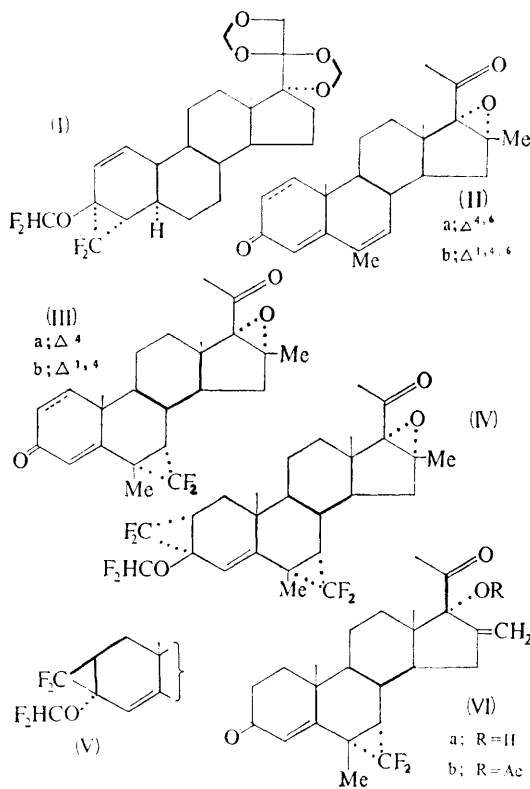
THE addition of difluoro- and dichloro-carbene to unsaturated steroids has been shown by Cross and his co-workers<sup>1</sup> to proceed readily. Subsequently, J. H. Fried and his colleagues<sup>2,3</sup> reported the formation of *gem*-difluorocyclopropanes by the addition of difluorocarbene to steroidal unsaturated ketones, and the formation of the difluoromethyl ether bis-adduct (I) from the corresponding  $\Delta^1$ -3-ketosteroid.<sup>2</sup> We now report the isolation of two new steroidal difluoromethyl ethers formed by incorporation of three CF<sub>2</sub> moieties into the steroid nucleus.

On reaction of (IIa) with excess of sodium chlorodifluoroacetate in triglyme at 170°, three compounds were formed which were separated by chromatography. The expected 6 $\alpha$ ,7 $\alpha$ -difluoromethylene adduct (IIIa)† was isolated in 33% yield: [m.p. 169–170°;  $[\alpha]_D^{25} + 98^\circ$ ;  $\lambda_{\max}$  (MeOH) 244.5 m $\mu$  ( $\epsilon$  15,100); n.m.r.  $\delta$  1.11 and 1.13 (C-10 and C-13 CH<sub>3</sub>), 1.40 (C-6 CH<sub>3</sub>, t,  $J_{HF}$  2.4 c./sec.), 6.03 (C-4 H); *M* calc.: 404, found: *M*<sup>+</sup> (*m/e*) 404].  $\alpha$ -Addition of difluorocarbene to the C-6 double bond is based on the absence of fluorine C-10 methyl proton coupling expected for the  $\beta$ -oriented difluoromethylene ring.<sup>4</sup>

Two additional products, each with three equivalents of difluorocarbene, were isolated in lesser amounts: (IV), 15% [m.p. 157–161°;  $[\alpha]_D^{25} + 58^\circ$ ; n.m.r.  $\delta$  1.06 (C-10 and C-13 CH<sub>3</sub>), 1.39 (C-6 CH<sub>3</sub>, t,  $J_{HF}$  2.5 c./sec.), 5.81 (C-4, H, b), 6.32 (C-3, OCHF<sub>2</sub>, q,  $J_{HF}$  72.8,  $J_{HF_2}$  74.0 c./sec.); *M* calc.: 504, found: *M*<sup>+</sup> (*m/e*) 504] and (V), 5% [m.p. 188–191°;  $[\alpha]_D^{25} - 20^\circ$ ;  $\lambda_{\max}$  (MeOH) 217 m $\mu$  ( $\epsilon$  8,330); n.m.r.  $\delta$  1.03 (C-13 CH<sub>3</sub>), 1.10 (C-10 CH<sub>3</sub>), 1.34 (C-6 CH<sub>3</sub>, t,  $J_{HF}$  2.5 c./sec.), 5.80 (C-4 H, b), 6.32 (C-3 OCHF<sub>2</sub>, q,  $J_{HF_1} = 73.5$ ,  $J_{HF_2} = 74.7$  c./sec.); *M* calc.: 504, found: *M*<sup>+</sup> (*m/e*) 504].

The assignments of structures (IV) and (V) to the minor products are supported by the following observations. The mass spectral data and the loss of  $\alpha\beta$ -unsaturated carbonyl absorptions in the i.r. and the u.v. spectra are consistent with the addition of three equivalents of difluorocarbene to the 3-keto-4,6-diene system. A characteristic n.m.r. signal is observed for the difluoromethyl ether

proton<sup>5</sup> in both products. The 2 $\alpha$ ,3 $\alpha$ ,6 $\alpha$ ,7 $\alpha$ -bis-difluoromethylene configuration is assigned to (IV) since a sharp signal is observed for the C-10 methyl proton resonance with no indication of long range fluorine coupling. However, the n.m.r. spectrum of the isomeric product (V) showed a broadened signal for the C-10 methyl protons ( $W_{\frac{1}{2}} = 2.5$ ) compared to that of the C-13 methyl proton band ( $W_{\frac{1}{2}} = 1.0$  c./sec.). This small but significant fluorine C-10 methyl proton coupling is consistent with a 2 $\beta$ ,3 $\beta$ -difluoromethylene configuration. Molecular models show that introduction of the C-4 double bond should decrease the magnitude of



† Satisfactory elementary analyses have been obtained for all new compounds. Rotations were determined in dioxane. N.m.r. spectra measured on a Varian A-60-A spectrometer in CDCl<sub>3</sub> solutions using Me<sub>4</sub>Si as internal standard. Mass spectra were determined on a CEC 21-103 spectrometer using a heated inlet system at a temperature of 200–230°. We thank Dr. T. Traubel for these measurements.

this coupling compared to the saturated ring system. Since the 6 $\beta$ ,7 $\beta$ -difluoromethylene configuration shows a relatively large fluorine C-10 methyl proton coupling ( $J_{\text{HF}}$  2—3 c./sec.),<sup>6</sup> (V) is favoured over the 2 $\alpha$ ,3 $\alpha$ ,6 $\beta$ ,7 $\beta$ - or the 2 $\beta$ ,3 $\beta$ ,6 $\beta$ ,7 $\beta$ -bisdifluoromethylene structures. The u.v. absorption maximum observed for (V) [ $\lambda_{\text{max}}$  (MeOH) 217 m $\mu$  ( $\epsilon$  8,330)] and not for (IV) further substantiates the relative stereochemical assignments of the A-ring cyclopropyl group. Kosower<sup>7</sup> has shown that interaction of a cyclopropyl and a carbonyl group, resulting in a bathochromic shift of the  $\pi \rightarrow \pi^*$  transition, is maximal when the plane of the ring and the  $p$ -orbitals of the carbonyl group are parallel. Dreiding models show that the plane of the 2,3-cyclopropyl group is approximately parallel to the plane of the  $p$ -orbitals of the C-4 double bond for (V) and close to perpendicular for (IV). A similar u.v. maximum [ $\lambda_{\text{max}}$  (C<sub>6</sub>H<sub>12</sub>) 218 m $\mu$  ( $\epsilon$  9,000)] was observed for the  $\alpha\beta$ -unsaturated cyclopropyl chromophore in 3 $\beta$ -acetoxy-5 $\alpha$ ,6 $\alpha$ -dibromomethylene-7,22-ergostadiene.<sup>8</sup>

Treatment of (IIb) under similar difluorocarbonylation conditions resulted in (IIIb) [m.p. 223—225.5°;  $\lambda_{\text{max}}$  (MeOH) 243 m $\mu$  ( $\epsilon$  14,950); n.m.r.  $\delta$  1.12 (C-13 CH<sub>3</sub>), 1.25 (C-10 CH<sub>3</sub>), 1.45 (C-6 CH<sub>3</sub>, t,  $J_{\text{HF}}$  2.4), 6.23 (C-2 H, q,  $J_{\text{H,H}_1}$  = 11,  $J_{\text{H,H}_2}$  = 2), 6.33 (C-4 H), 6.95 (C-1 H, d,  $J_{\text{H,H}_1}$  = 11 c./sec.);  $M$  calc.: 402, found:  $M^+$  ( $m/e$ ) 402].

Reaction of (IIIa) with catalytic amounts of sulphuric acid in dioxan<sup>9</sup> gave (VIa), which on acetylation with acetic acid, trifluoroacetic anhydride, and toluene- $p$ -sulphonic acid at room temperature<sup>10</sup> yielded (VIb) [m.p. 205—207°;  $[\alpha]_{\text{D}}^{25}$  -60°;  $\lambda_{\text{max}}$  (MeOH) 244.5 m $\mu$  ( $\epsilon$  14,100); n.m.r.  $\delta$  0.74 (C-13 CH<sub>3</sub>), 1.14 (C-10 CH<sub>3</sub>), 1.42 (C-6 CH<sub>3</sub>, t,  $J_{\text{HF}}$  = 2.4 c./sec.), 6.04 (C-4 H)]. (VIb) was also obtained when 17 $\alpha$ -acetoxy-6-methyl-16-methylene-4,6-pregnadien-3,20-dione was treated with an excess of sodium chlorodifluoroacetate in triglyme at 200°.

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